

Strong Spin-Lattice Coupling Through Oxygen Octahedral Rotation in Divalent Europium Perovskites

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First-principles calculations reveal that in divalent europium perovskites EuMO_3 ($M = \text{Ti, Zr, and Hf}$), antiferromagnetic superexchange interactions via nd states of the B-site M cations ($n = 3, 4$, and 5 , respectively) are enhanced by rotations of the MO_6 octahedra. The octahedral rotations involved in a structural change from cubic $Pm\bar{3}m$ to orthorhombic $Pbnm$ structures not only reduce energy gaps between the Eu $4f$ and M nd bands but also point the M nd orbitals at the Eu sites, leading to a significant overlap between the M nd and Eu $4f$ orbitals. These results reveal that the octahedral rotations are indispensable for antiferromagnetic ordering observed for EuZrO_3 and EuHfO_3 , and put these perovskites into a class of materials exhibiting a novel type of strong coupling between their magnetism and octahedral rotations.

1. Introduction

ABO_3 perovskite oxides often present structural distortions associated with rotations of the BO_6 octahedra, leading to lower symmetry than an ideal $Pm\bar{3}m$ perovskite represented by $a^0a^0a^0$ with Glazer notation (Figure 1a).^[1] Focusing on the ubiquitous distortions, there have been symmetry arguments on macroscopic effects related to the octahedral rotations.^[2–7] For example, a new type of symmetry operation termed a roto operation, which locally reverses the sense of the oxygen octahedral rotations, has been introduced to understand physical quantities coupled to the octahedral rotations such as rotomagnetism, in which magnetization is induced by octahedral rotations.^[2] In addition, a path to design of ferroelectric (or multiferroic) materials by actively utilizing the oxygen octahedral rotations through a hybrid improper mechanism has been suggested.^[3,4,6] In this mechanism, a polar distortion is induced by a combination

of two kinds of octahedral rotations without introducing second-order Jahn–Teller active cations, i.e., lone-pair cations such as Pb^{2+} and d^0 transition metal cations such as Ti^{4+} , to the A and B sites of perovskites, respectively.^[6]

In addition to such symmetry arguments that predict the occurrence of as-yet-unestablished effects from a macroscopic viewpoint,^[2–7] it is important in the material design to microscopically understand the origin and magnitude of the effects. Large rotomagnetic effects and robust multiferroic coupling through the hybrid improper mechanism are expected if magnetic interactions are induced and/or drastically changed by the octahedral

rotations. What comes immediately to mind as a mechanism connecting magnetism to the octahedral rotations could be a superexchange interaction via an oxide ion^[8,9] since their strength strongly depends on metal-oxygen-metal bond angles and hence the octahedral rotations. Materials whose magnetic interactions are intimately linked to the octahedral rotations via a novel mechanism are demanded to develop new functionalities such as the rotomagnetic effect.

Recently, we have presented first-principles calculations showing that antiferromagnetic (AFM) superexchange interactions via M nd states play a predominant role in AFM ordering of Eu^{2+} $4f$ spins ($S = 7/2$) for divalent europium perovskites EuMO_3 ($M = \text{Ti, Zr, and Hf}$, and $n = 3, 4$, and 5 , respectively).^[11] A strong spin-lattice coupling observed in EuTiO_3 , including a large enhancement of the dielectric constant by a magnetic field^[12] and ferromagnetism induced by a tensile strain,^[13,14] a lattice expansion,^[15–17] and an application of an electric field,^[18] is explained by a competition between the AFM superexchange interaction via the Ti $3d$ states and ferromagnetic (FM) indirect exchange interaction via the Eu $5d$ states.^[19,20] Since the M nd states are strongly affected by the rotations of the MO_6 octahedra, the strength of superexchange interactions are expected to depend on the octahedral rotations.^[21,22]

In this study, we perform systematic first-principles calculations on the magnetic interactions and electronic structure of the Eu^{2+} perovskites for prototypal cubic ($Pm\bar{3}m$) and orthorhombic ($Pbnm$) models (Figure 1), the latter of which is represented by $a^-a^-c^+$ and experimentally observed for EuZrO_3 and EuHfO_3 at room temperature.^[21] As a result, we find that the AFM superexchange interaction is strengthened by rotations of the MO_6 octahedra involved in the cubic to orthorhombic

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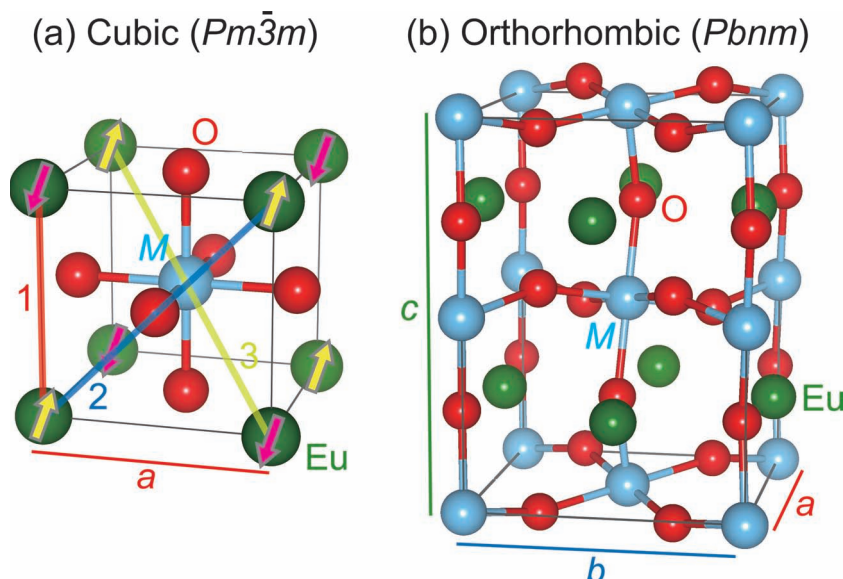


Figure 1. Schematics of a) cubic perovskite structure with space group $Pm\bar{3}m$ and b) orthorhombic structure with space group $Pbnm$. The arrows in (a) schematically indicate a G-type magnetic structure. The numbers 1, 2, and 3 denote the first, second, and third nearest-neighbor Eu-ion pairs, respectively. The crystal structures are visualized using VESTA code.^[10]

structural change. Our calculations reveal that the magnetic ground state of EuZrO_3 and EuHfO_3 is changed from FM to AFM as their structure is varied from cubic to orthorhombic. An analysis of charge density shows that in EuZrO_3 , overlap between the Eu 4f and Zr 4d states becomes larger with the structural change; the ZrO_6 octahedral rotations lower the site symmetry of Zr^{4+} , causing anisotropic development of the lobes of the 4d states of Zr^{4+} toward the Eu sites. The enhancement of the superexchange interactions is explained based on a third-order perturbation. The superexchange interactions are strengthened due to the enhancement of the overlap integral between the Eu 4f and Zr 4d states as well as due to a decrease in energy gaps between the Eu 4f and Zr 4d states. Such enhancement of the AFM superexchange interactions occurs in EuTiO_3 and EuHfO_3 as well. Our results indicate that these perovskites possess a novel type of strong coupling between the magnetic properties and octahedral rotations.

2. Computational Methods

The calculations were performed using the projector augmented-wave (PAW) method^[23] and the HSE06 hybrid functional^[24–26] as implemented in the VASP code.^[27–29] The HSE06 functional adopts the replacement of one-quarter of the exchange energy in the Perdew–Burke–Ernzerhof generalized gradient approximation (PBE-GGA)^[30] with the nonlocal Fock exchange, and a screening parameter of 0.208 \AA^{-1} . It has been shown that the hybrid Hartree–Fock density functional approach describes the electronic structure for a variety of molecules and solids more precisely than those obtained from the local and semilocal functional.^[11,24–26,28,29,31–39] The calculation conditions have been described in detail in our previous reports.^[11,21]

We mainly consider structural models possessing a space group of $Pm\bar{3}m$ and $Pbnm$ with various cell volumes, which are referred to as cubic and orthorhombic models below, respectively. Cell shapes and internal coordinates were relaxed at any volume under the constraints of the symmetry of the unit cells. Tetragonal $\sqrt{2} \times \sqrt{2} \times 2$ supercells with four formula units were used for the cubic model in order to represent four magnetic configurations i.e., AFM A-, C-, G-, and FM F-types,^[15,16] the total energies of which are necessary to estimate the first, second, and third nearest-neighbor (NN) exchange constants, J_1 , J_2 , and J_3 , respectively, shown in Figure 1a. We confirmed that the ratio of the tetragonal lattice constants a to c approximately remained 1 to $\sqrt{2}$ at any volume. For the orthorhombic model, we use the unit cells illustrated in Figure 1b. The A-type ordering is assumed to have FM layers within the ab plane stacking antiferromagnetically along the c axis, and the C-type ordering has FM chains along the c axis that couple to the neighboring chains antiferromagnetically. J_1 , J_2 , and J_3 are defined as in the case of the

cubic model by regarding the lattice composed of Eu^{2+} ions as a simple cubic. Actually, there are three kinds of J_1 , four kinds of J_2 , and four kinds of J_3 . In estimating the exchange constants by using total energies of the A- and C-types defined above, simply-averaged values are estimated for J_1 and J_3 . However, estimated for J_2 are averaged values of the exchange constants between the second NN spins in the neighboring ab layers while the exchange constants between the second NN spins within the same ab layer are not included due to the model shape.

The first, second, and third NN exchange constants were calculated by mapping the total-energy difference between different magnetic configurations onto the Heisenberg spin Hamiltonian $H = -2 \sum_{i,j} J^{ij} \mathbf{S}_i \cdot \mathbf{S}_j$, where J^{ij} is the exchange constant of magnetic interactions between the Eu 4f spins at the i and j sites and \mathbf{S}_i is the spin vector at the i site.^[11,13,15,16,21] The total energies per unit cell for A-, C-, F, and G-types, respectively, are represented by the exchange constants as follows:

$$E_A = E_0 + 2|S|^2(-4J_1 + 8J_2 + 16J_3) \quad (1)$$

$$E_C = E_0 + 2|S|^2(4J_1 + 8J_2 - 16J_3) \quad (2)$$

$$E_F = E_0 + 2|S|^2(-12J_1 - 24J_2 - 16J_3) \quad (3)$$

$$E_G = E_0 + 2|S|^2(12J_1 - 24J_2 + 16J_3) \quad (4)$$

where $|S|^2 = S(S+1)$ and $S = 7/2$ for Eu^{2+} . The values of J_1 , J_2 , and J_3 were obtained by substituting calculated total energies into the left-hand sides of Equation (1) to Equation (4) and then solving the simultaneous equations.

In order to estimate stabilization energies accompanied by structural distortions related to the MO_6 octahedral rotations,

the calculations were also performed for four other typical perovskite structures with different tilting patterns, i.e., $I4/mcm$ ($a^0a^0c^-$), $P4/mbm$ ($a^0a^0c^+$), $Cmmm$ ($a^-b^0c^+$), and $Ibmm$ ($a^-a^-c^0$) structures.^[40] The cell parameters and atomic coordinates were relaxed under the constraints of the symmetry. As for EuTiO_3 with an $I4/mcm$ structure, which is experimentally observed at low temperatures,^[41–45] we calculated total-energy differences between G - and F -type magnetic configurations as a function of cell volume for comparison with experiments. This structural model is denoted as a tetragonal model below.

3. Results and Discussion

We begin with reviewing fundamental properties of the Eu^{2+} perovskites including structural, magnetic, and optical properties investigated experimentally and theoretically so far. EuTiO_3 has a cubic $Pm\bar{3}m$ structure at room temperature, while EuZrO_3 and EuHfO_3 have orthorhombic $Pbnm$ structures.^[12,21,46–48] As for EuTiO_3 , the occurrence of antiferro-distortive transition from cubic $Pm\bar{3}m$ to tetragonal $I4/mcm$ structure has been recently indicated by a specific heat measurement,^[41] variable-temperature powder X-ray diffraction,^[42,43] electron diffraction,^[43] and first-principles calculations.^[44,45] The lattice constants are well reproduced by the calculations.^[21] We compare in Figure 2 the total energies of EuMO_3 ($M = \text{Ti, Zr, and Hf}$) with different octahedral tilting patterns to that of the cubic $Pm\bar{3}m$ structure taken as a reference.^[40] For EuZrO_3 and EuHfO_3 , the lowest energy is obtained for the $Pbnm$ structure in agreement with the experimental results. On the other hand, for EuTiO_3 , the structures with $I4/mcm$ and $Ibmm$ symmetry are more stable than the $Pm\bar{3}m$ structure, and they are nearly degenerate. The $P4/mbm$, $Cmmm$, and $Pbnm$ structures relax to the $Pm\bar{3}m$, $I4/mcm$, and $Ibmm$ structures, respectively. These results agree well with the recent results of GGA+ U calculations.^[45] Hereinafter, we refer $Ibmm$ structure for EuTiO_3 and $Pbnm$ structure for EuZrO_3 and EuHfO_3 as orthorhombic structures for simplicity.

The optical band gaps are estimated to be 0.8, 2.4, and 2.7 eV for EuTiO_3 , EuZrO_3 , and EuHfO_3 , respectively, from diffuse

reflectance spectra measured at room temperature.^[21] Figure 3 shows site-projected density of states (DOS) of the perovskites for both cubic and orthorhombic models. Band gaps, E_g , which are also shown in Figure 3, are in good agreement with the experimental optical band gaps. The highest occupied bands are mainly composed of Eu 4f states, and have a narrow width, indicating the localized nature of the Eu 4f electrons. It should be noted that M nd components ($M = \text{Ti, Zr, and Hf}$ and $n = 3, 4, \text{ and } 5$, respectively) as well as O 2p components are also observed in the energy region of the Eu 4f bands. The O 2p bands lie below the Eu 4f bands. The lowest unoccupied bands, on the other hand, chiefly consists of the M nd states. The Eu 5d states form broad bands that are located above the M nd states.

Paramagnetic to AFM transitions are observed at $T_N = 5.5, 4.1, \text{ and } 3.9 \text{ K}$ for EuTiO_3 , EuZrO_3 , and EuHfO_3 , respectively. Powder neutron diffraction study indicates that EuTiO_3 has a G -type magnetic configuration below T_N .^[49] Negative values of J_1 and positive values of J_2 are derived from Weiss temperature, θ_W , and Néel temperature, T_N , based on the molecular-field theory.^[12,21,47,48] For tetragonal EuTiO_3 and orthorhombic EuZrO_3 and EuHfO_3 as well as cubic EuTiO_3 , the calculated total-energy differences between G - and F -type magnetic configurations, $E_G - E_F = [-2|S|^2(24J_1 + 32J_2)]$, which correspond to an inter-sublattice interaction in the G -type AFM configuration (see the arrows in Figure 1a), are plotted as a function of cell volume in Figure 4a. The calculations reproduce AFM ground states for the perovskites at their equilibrium cell volumes. The energy difference can be experimentally derived from θ_W and T_N as $E_G - E_F = 6k_B(\theta_W - T_N)$ since according to the molecular-field theory, $\theta_W = 2|S|^2/3k_B(6J_1 + 12J_2 + 8J_3)$ and $T_N = 2|S|^2/3k_B(-6J_1 + 12J_2 - 8J_3)$ within the third NN interactions. The experimental values are also plotted in Figure 4a against the experimental cell volumes. These values are well reproduced by the calculations. An increase in cell volume leads to switching of the magnetic ground state from AFM to FM for both cubic and tetragonal EuTiO_3 , which is in good agreement with the experimental results that FM behavior is observed for epitaxial EuTiO_3 thin films with a c -axis elongation^[17] and a biaxial tensile strain.^[14] This behavior is interpreted in terms

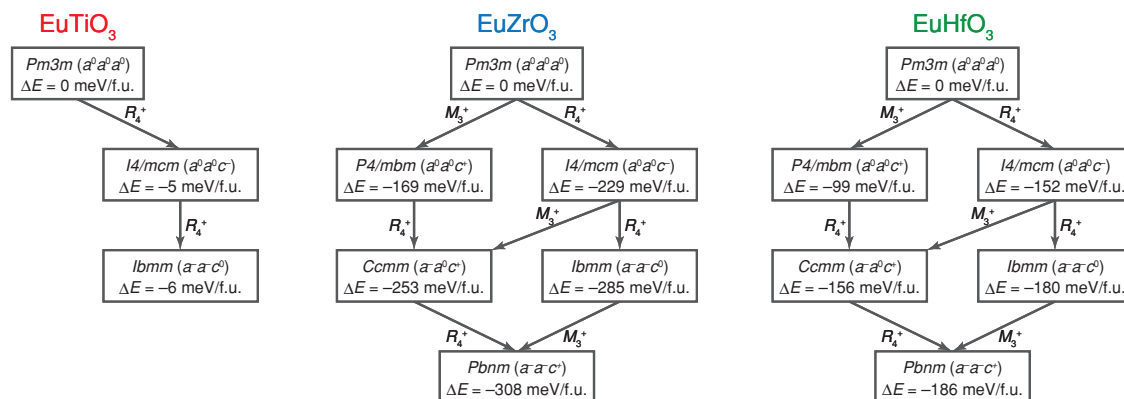
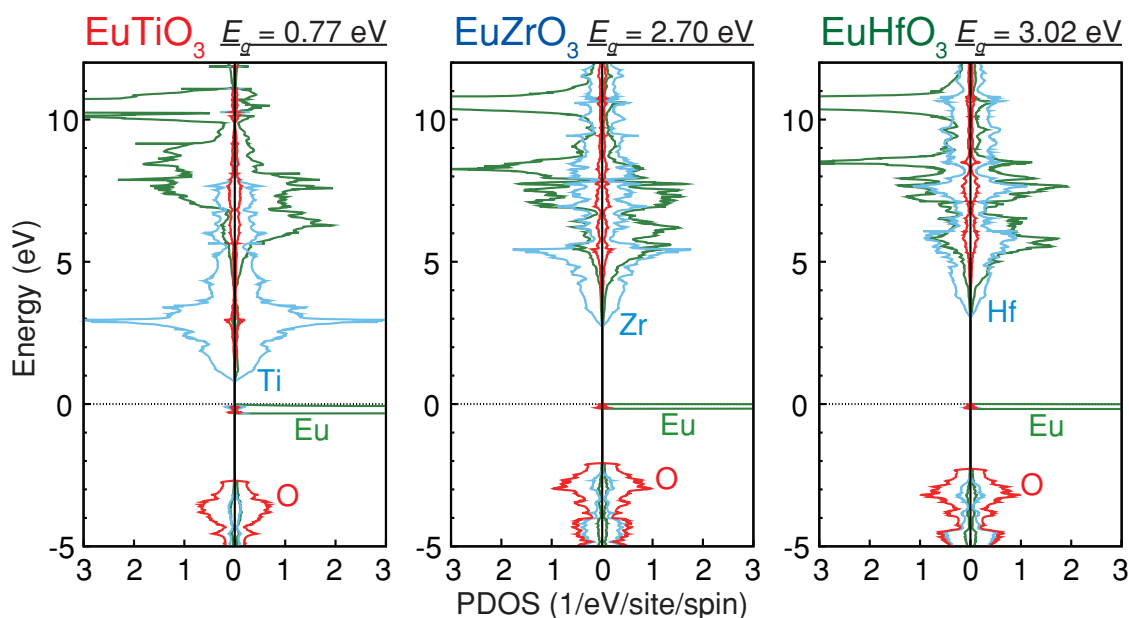


Figure 2. Total energies of EuMO_3 ($M = \text{Ti, Zr, and Hf}$) with different octahedral tilting patterns relative to the energy of the cubic $Pm\bar{3}m$ ($a^0a^0a^0$) structure for the G -type magnetic configuration. The types of additional octahedral rotations are denoted by M_3^+ (in phase) and R_4^+ (out of phase).

Cubic



Orthorhombic

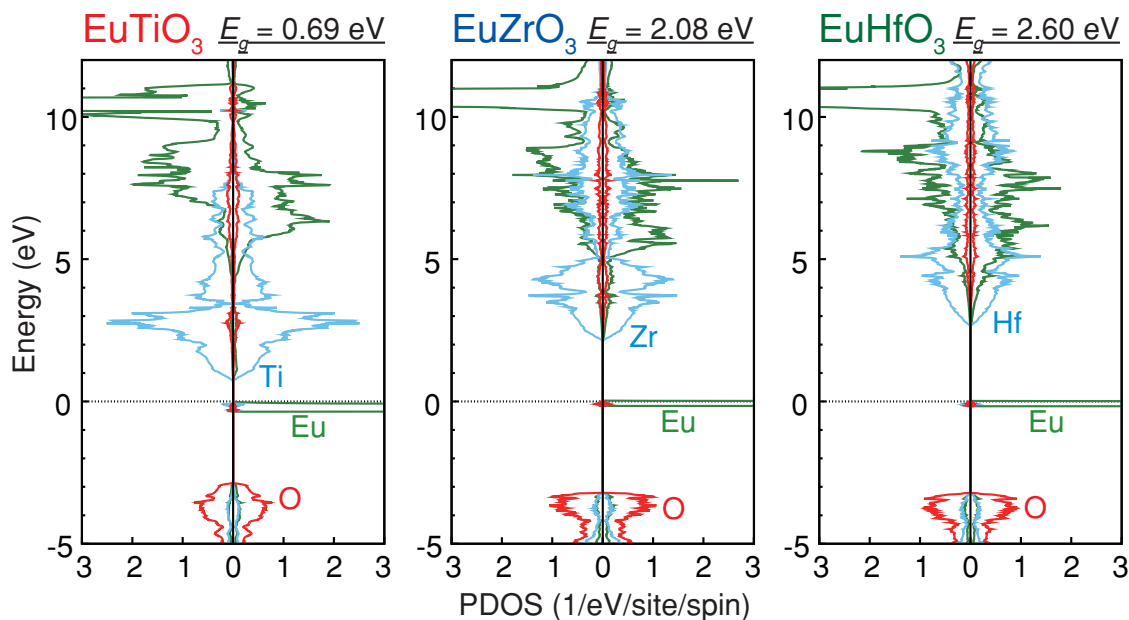


Figure 3. Site-projected density of states for EuMO_3 [$M = \text{Ti}$ (left), Zr (center), and Hf (right)] in cubic (top) and orthorhombic (bottom) models. The G -type magnetic configurations are adopted. The zero energies are set to the highest occupied states. Band gaps (E_g) estimated as a difference between the eigenvalues for the lowest unoccupied and highest occupied states are also shown.

of competition between the AFM superexchange interactions via the Ti 3d states and the FM indirect exchange interactions via the Eu 5d states in the inter-sublattice interaction.^[11] In orthorhombic EuZrO_3 and EuHfO_3 , the AFM superexchange interactions via the Zr 4d and Hf 5d states, respectively, are so stronger than the FM exchange interaction in the calculated volume range that no sign reversal of J_1 occurs and J_1 decreases monotonously with a decrease in cell volume. Figure 4a

also shows the cell-volume dependence of $E_G - E_F$ for as-yet-unsynthesized cubic EuSiO_3 for comparison. It has a negative slope against cell volume in sharp contrast to the above three perovskites because the FM exchange interaction is predominant for Si due to the absence of d states that mediate the AFM superexchange interactions.^[11] A charge-density difference between two compounds is helpful in understanding the difference in their chemical bonding.^[39] The charge-density

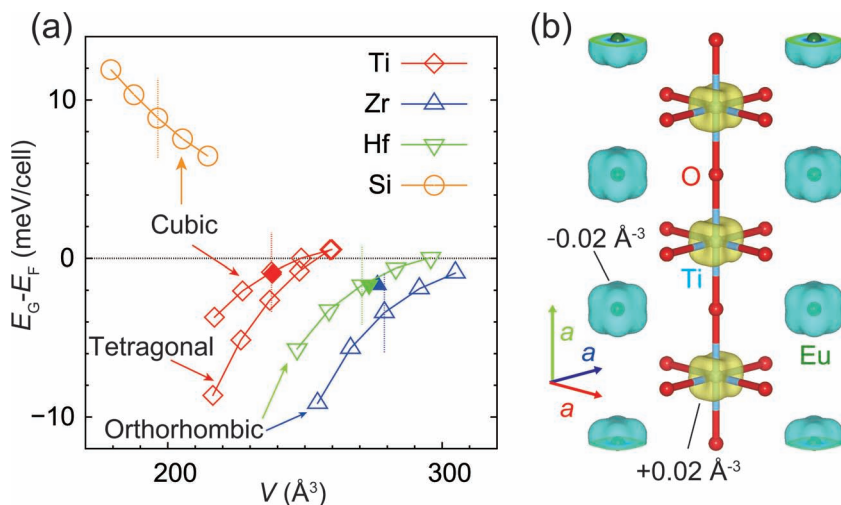


Figure 4. a) Cell-volume dependence of the total-energy difference between *G*-type antiferromagnetic and ferromagnetic states for EuTiO₃ for the cubic and tetragonal (*I4/mcm*) models, EuMO₃ with *M* = Zr and Hf for the orthorhombic models, and EuSiO₃ for the cubic model. The broken vertical lines indicate the equilibrium volumes. The energy differences estimated experimentally are plotted against the experimental cell volume as indicated by the filled marks.^[21] b) Difference in charge density integrated within the energy region of Eu 4f bands between cubic EuTiO₃ and EuSiO₃. The structure with the averaged lattice constant between two optimized cubic structures was used. The yellow (+0.02 Å⁻³) and blue (-0.02 Å⁻³) isosurface means excess and deficiency of charge in EuTiO₃ relative to EuSiO₃. The crystal structures and isosurfaces are visualized using VESTA code.^[10]

difference for the Eu 4f bands between cubic EuTiO₃ and EuSiO₃ are depicted in Figure 4b. The charge density around the Eu and Ti sites is decreased and increased, respectively, in EuTiO₃ when compared to EuSiO₃ while that around the O sites is similar. This result supports the significance of contribution of Ti 3d states in the Eu 4f bands. This is also seen in the site-projected DOS shown in Figure 3 as a mixing of the Ti 3d states in the Eu 4f bands.

Thus, the calculations well reproduce the fundamental properties such as lattice constants, band gaps, and magnetic structures observed experimentally. However, if the relationship between the electronic structure and exchange constants across the perovskites is carefully considered, we face seemingly puzzling behavior. According to an analysis based on the molecular-orbital theory, the superexchange interactions between the Eu 4f spins via the *M* *nd* states are approximately represented as a third-order perturbation from an atomic orbital basis as follows:

$$J_{SE} \propto \frac{\sigma_{df}^4}{(\Delta E)^2} \quad (5)$$

where σ_{df} and ΔE are an overlap integral and an energy gap between the Eu 4f and *M* *nd* orbitals.^[50] This equation indicates that the AFM superexchange interaction becomes weaker rapidly with an increase in ΔE if σ_{df} is constant. E_g approximately corresponds to ΔE . The site-projected DOS shown in Figure 3 shows that E_g increases from cubic EuTiO₃ to orthorhombic EuZrO₃ and EuHfO₃. However, the AFM interactions between the sublattices are, in fact, stronger in orthorhombic EuZrO₃ and EuHfO₃ than in cubic EuTiO₃ (see Figure 4a), in contrast to

a trend expected from Equation (5) in terms of ΔE . This implies that another factor than ΔE , i.e., σ_{df} , could be significantly varied between the cubic and orthorhombic structures. It should be also noted that the AFM inter-sublattice interactions are significantly stronger for tetragonal EuTiO₃ than for cubic EuTiO₃ although E_g is almost the same for the both models ($E_g = 0.73$ eV for tetragonal EuTiO₃; data is not shown), implying that the small octahedral-rotation angle around the *c* axis (6.35°) in the tetragonal model brings about the enhancement of the interactions, probably, due to an increase in σ_{df} . Therefore, we concentrate on discussion of the difference in the magnetic interactions and electronic structure between the cubic and orthorhombic models below.

The calculated values of J_1 , J_2 , and J_3 are plotted in Figure 5 as a function of cell volume for the perovskites with the cubic and orthorhombic models. The magnitude of J_3 is much smaller than those of J_1 and J_2 for any perovskites with any structural models. Therefore, the cell-volume dependence of J_1 is similar to that of $E_G - E_F = 2[S]^2(24J_1 + 32J_3)$ shown in Figure 4a for cubic EuTiO₃ and orthorhombic EuZrO₃ and EuHfO₃. Interest-

ingly, EuZrO₃ and EuHfO₃ with the cubic models show positive values of J_1 and hence the FM ground state in the whole volume range in sharp contrast to the case of the orthorhombic models, while J_2 is almost independent of the structural models. In addition, the slopes of J_1 against cell volume are negative for the cubic model contrary to those for the orthorhombic model. The behavior of J_1 for the cubic models across the perovskites can be explained in terms of ΔE ; the AFM superexchange interactions are much weaker for EuZrO₃ and EuHfO₃ than for EuTiO₃ due to the larger ΔE if σ_{df} is almost same. For any perovskites, changing from cubic to orthorhombic, J_1 decreases significantly, indicating that the orthorhombic *Pbnm* structures with the MO₆ octahedral rotations favor the AFM coupling between the first NN Eu 4f spins compared to the cubic *Pm3m* structures.

Figures 6a,b illustrate the MO₆ octahedral rotation patterns in the orthorhombic *Pbnm* structure (*a*⁻*a*⁻*c*⁺). Here, the two rotation angles, θ^a and θ^c , are defined as shown in Figures 6a,b. For orthorhombic EuZrO₃ and EuHfO₃ at the equilibrium volume, (θ^a , θ^c) are (9.2°, 8.4°) and (8.2°, 7.4°), respectively. These values are slightly larger than the experimental values obtained from synchrotron X-ray diffraction at room temperature, i.e., (7.13°, 6.64°) and (7.18°, 5.60°) for EuZrO₃ and EuHfO₃, respectively.^[21] Both θ^a and θ^c increase with a decrease in cell volume. Such octahedral rotations could enhance the AFM superexchange interaction. As mentioned above, the *Pbnm* structure is unstable and relaxes to *Ibmm* structure (*a*⁻*a*⁻*c*⁰) for EuTiO₃. θ^a becomes larger as cell volume decreases, while θ^c is zero within the numerical error. Although the rotation angles are smaller in EuTiO₃ than those in EuZrO₃ and EuHfO₃, the AFM superexchange interaction is significantly enhanced as shown in Figure 5.

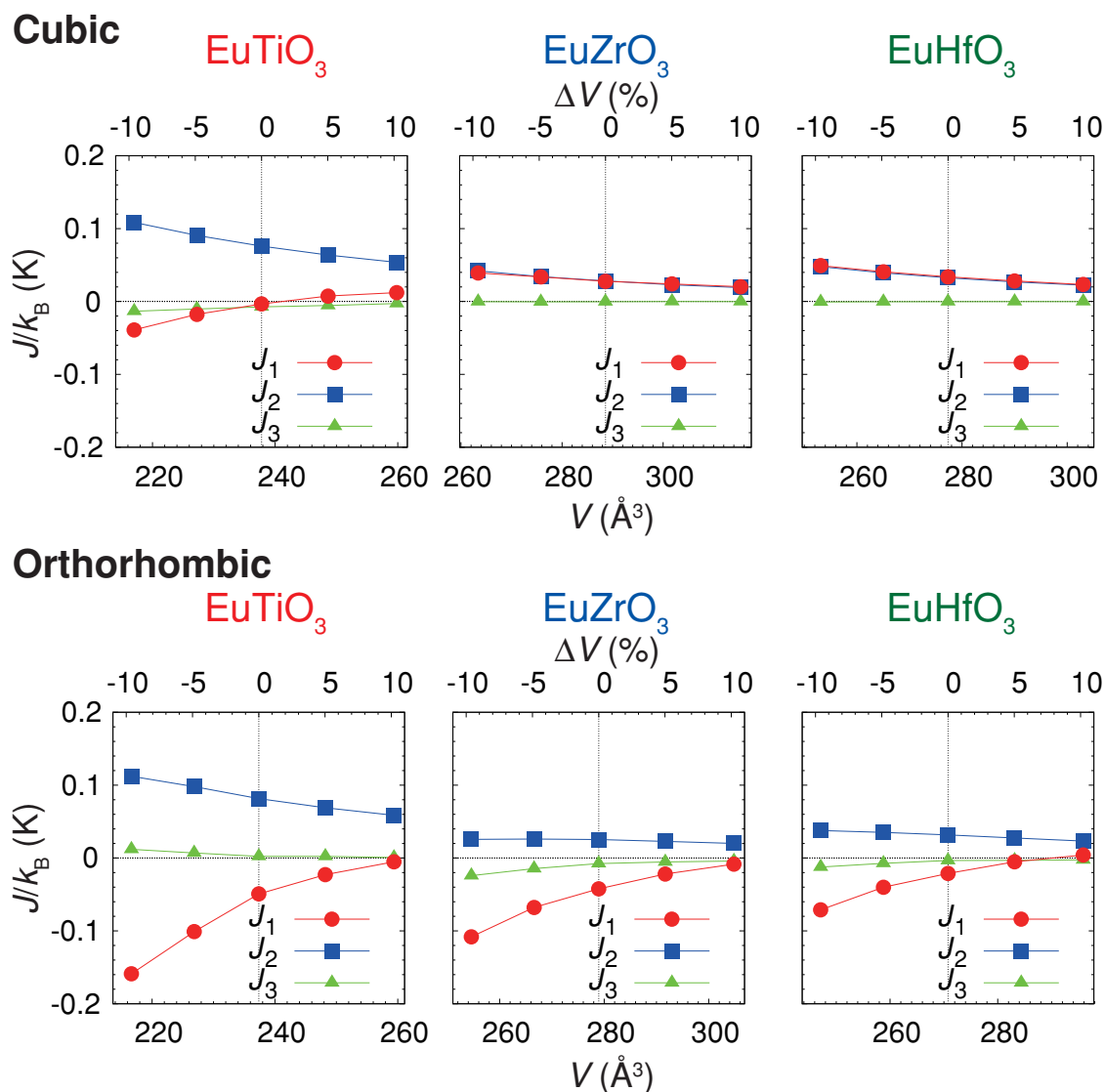


Figure 5. Cell-volume dependence of J_1 , J_2 , and J_3 for EuMO_3 [$M = \text{Ti}$ (left), Zr (center), and Hf (right)] with cubic (top) and orthorhombic (bottom) models. $\Delta V = 0$ indicates the equilibrium volume. Positive and negative values indicate FM and AFM interactions, respectively.

We now discuss the microscopic origin of the enhancement of the AFM interactions by the octahedral rotations. As seen in Equation (5), the important factors in the AFM superexchange interactions are an energy gap (ΔE) and an overlap integral (σ_{df}) between the Eu 4f and M nd orbitals. The site-projected DOS and E_g values are shown for the perovskites with the cubic and orthorhombic models in Figure 3. Within the cubic models, the fact that J_1 is much lower in EuTiO_3 than in EuZrO_3 and EuHfO_3 can be interpreted in terms of the difference in ΔE , as mentioned above. The E_g values decrease by at most 25% when the structures are changed from cubic to orthorhombic. When one compares each compound with cubic and orthorhombic structures, the enhancement of the AFM superexchange interactions accompanied by the structural change seems to be difficult to explain only in terms of such small decreases in ΔE , in particular, for EuZrO_3 and EuHfO_3 . In addition, when

comparing orthorhombic EuTiO_3 and EuZrO_3 , J_1 is similar to each other at the equilibrium volume while E_g is much larger for EuZrO_3 (2.10 eV) than that for EuTiO_3 (0.69 eV). These facts suggest the importance of another factor for the AFM superexchange interaction, i.e., σ_{df} . Therefore, it is necessary to consider a difference in the overlap of the M d orbitals with the Eu 4f orbitals between the cubic and orthorhombic structures to understand the enhancement of the AFM interactions by the octahedral rotations.

To focus on the overlap between the Eu 4f and M d orbitals, charge-density isosurfaces of the Eu 4f bands are illustrated as a function of cell volume for cubic EuTiO_3 and cubic and orthorhombic EuZrO_3 in Figure 7. For cubic EuTiO_3 (Figure 7a), a non-negligible amount of charge density is recognized around the Ti and O sites as well as the Eu sites, indicating a hybridization of the Eu 4f states with the Ti 3d and O 2p states.

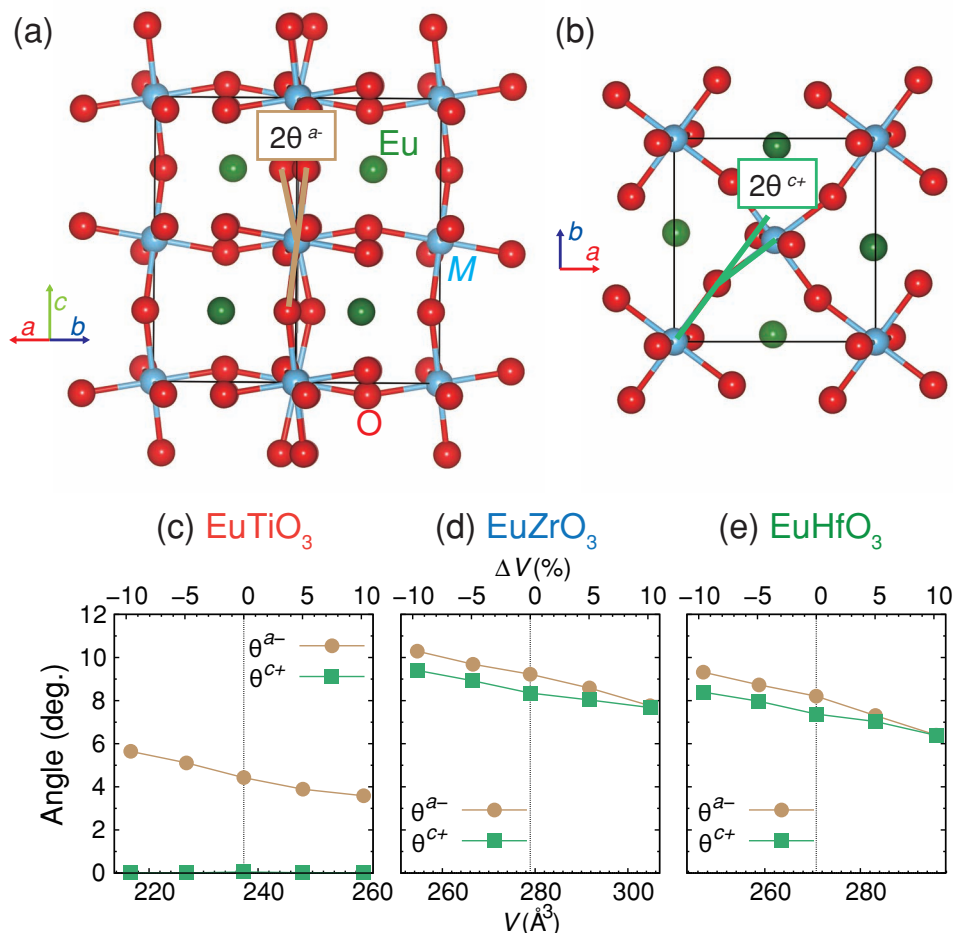


Figure 6. Schematics of MO_6 octahedral rotations along the direction a) perpendicular to and b) parallel to the c axis in an orthorhombic $Pbnm$ structure ($a^-a^-c^+$). The rotation angles θ^{a-} and θ^{c+} are defined in (a) and (b), respectively. Cell-volume dependence of θ^{a-} and θ^{c+} for orthorhombic $EuMO_3$ for $M = c)$ Ti, d) Zr, and e) Hf with G-type magnetic configurations. $\Delta V = 0$ indicates the equilibrium volume.

The charge density around the Ti sites reflects its site symmetry ($m\bar{3}m$), and thereby the Ti 3d orbitals overlap equally with the Eu 4f orbitals at the eight equivalent Eu sites. A decrease in cell volume leads to an increase in the charge density around the Ti sites, indicating the enhancement of the hybridization of the Eu 4f and Ti 3d states and hence the AFM superexchange interaction via the Ti 3d states. This result corresponds to a decrease in J_1 with lowering cell volume (see Figure 5). On the other hand, no charge-density isosurfaces around the Zr sites are recognized for cubic $EuZrO_3$ with any cell volume. This indicates that the hybridization between the Eu 4f and M d states is much weaker in cubic $EuZrO_3$ than in cubic $EuTiO_3$. Therefore, the AFM superexchange interaction via the Zr 4d states is overwhelmed by the competing FM exchange interaction via the Eu 5d states for J_1 in cubic $EuZrO_3$, resulting in the FM ground states. In the case of orthorhombic $EuZrO_3$, one can see anisotropic development of charge density around the Zr ions with site symmetry of $\bar{1}$. The lowering of site symmetry accompanied by the $a^-a^-c^+$ octahedral rotations allows the anisotropic spreading of the Zr 4d orbital. This clearly corroborates that the AFM superexchange interaction via the Zr 4d states

is enhanced in orthorhombic $EuZrO_3$ as compared to cubic $EuZrO_3$. It should be noted that although the lobes of the Zr 4d orbitals extend, in particular, toward the two Eu ions that are the nearest to the Zr ions, the distance between the Eu and Zr ions is not the most crucial factor for the AFM superexchange interaction; we have confirmed that a considerable amount of charge density is discerned around the Zr sites in orthorhombic $EuZrO_3$ with $\Delta V = +9.3\%$ but not in cubic $EuZrO_3$ with $\Delta V = -8.7\%$, in which the equivalent Eu-Zr distance is smaller than the smallest Eu-Zr distance in the former (data not shown). The site-symmetry lowering by the octahedral rotations is the most essential factor for the enhancement of the interaction between the Eu 4f and Zr 4d states.

4. Conclusions

Using first-principles calculations, we have investigated the electronic structure, magnetism, and their relationship for $EuMO_3$ ($M = \text{Ti, Zr, and Hf}$) with the cubic $Pm\bar{3}m$ and orthorhombic $Pbnm$ models. The MO_6 octahedral rotations accompanied by

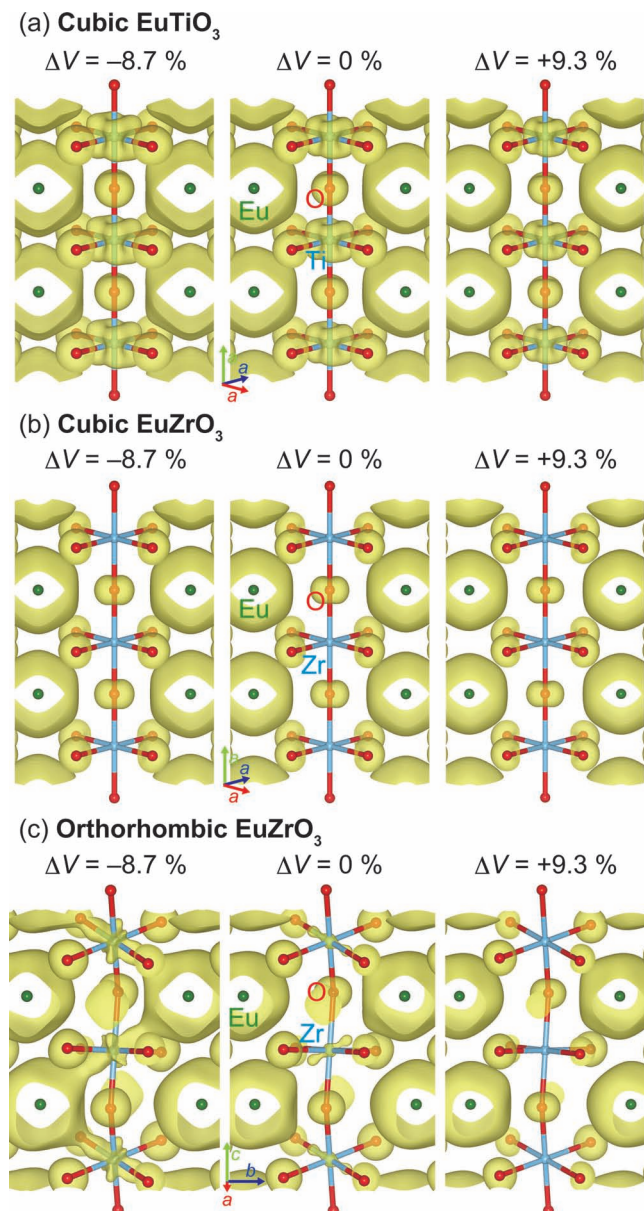


Figure 7. Cell-volume dependence of isosurfaces of charge density integrated within the energy region of the Eu 4f bands for G-type a) cubic EuTiO_3 and b) cubic and c) orthorhombic EuZrO_3 . The isosurfaces are depicted at 0.005 \AA^{-3} .

the structural change from cubic to orthorhombic lead to the drastic enhancement of the AFM superexchange interactions between the Eu 4f spins via the $M nd$ states ($M = \text{Ti, Zr, and Hf}$, and $n = 3, 4$, and 5). The octahedral rotations not only reduce energy gaps between the Eu 4f and $M nd$ states but also facilitate a spatial overlap between the Eu 4f and $M d$ orbitals. One of the remarkable results is that the octahedral rotations alter the magnetic ground states for EuZrO_3 and EuHfO_3 : the magnetic ground state is switched from FM into AFM when the structure is changed from cubic to orthorhombic. In other words, the octahedral rotations are indispensable for AFM ordering of the Eu 4f spins observed experimentally in EuZrO_3 and EuHfO_3 .

Thus, the present results reveal that these perovskites belong to a class of materials exhibiting a novel type of strong correlation between their magnetism and octahedral rotations. In closing, although investigations on strain effects are outside of the scope of this paper, the magnetic and dielectric properties of EuZrO_3 and EuHfO_3 epitaxial thin films are very fascinating in that they could be multiferroic (FM and ferroelectric) as with EuTiO_3 ,^[13] by using strain engineering so as to suppress the octahedral rotations and induce a polar distortion.^[7,40,51,52]

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